	, ;	•	ķ 4	. 1	3.1	i
(U)						
Y CLASSIFICA	OIT.	Ν	OF	THIS	PAC	ìΕ

								Form Approved OMB No. 0704-0188	
;	ORT SECURITY CLASSIFICATION ELECTE 16. RESTRICTIVE MARKINGS NA								
ガ	JRITY CLASSIFICATION AUTHORY SEP 2 6 1988 3. DISTRIBUTION/AVAILABILITY OF REPORT								
,	LASSIFICATION / DOWNG	Distribution Unlimited							
1	NA	DEDORT ANIMAGE	2/5)						
	DRMING ORGANIZATION REPORT NUMBER(S)		S. MONITORING ORGANIZATION REPORT NUMBER(S) NA						
.	Rice Univers		e d'atamente de la compa	<u></u>					
7	Rice Univers		6b. OFFICE SYMBOL (If applicable)	7a. NAME OF MONITORING ORGANIZATION Office of Naval Research					
•		<u>-</u>	NA				searc	n	
	Physics Don:			7b. ADDRESS (City, State, and ZIP Code)					
l	Physics Department P.O. Box 198			800 N. Quincy Street Arlington, VA 22217-5000					
	Houston, TX	77251							
	ME OF FUNDING / SPONS SANIZATION	OR:NG	8b OFFICE SYMBOL (If applicable)	9 PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER					
	ice of Naval		ONR	1	4-86-K-00				
8c ADD	RESS (City, State, and ZIF			10 SOURCE OF F	UNDING NUMBER	S TASK		WORK UNIT	
	800 N. Quino		7 5000	ELEMENT NO	NO.	NO.		ACCESSION NO.	
	Arlington, N		7-5000	61153N	RR04108	4414	4704		
11. TITL (U)	E (Include Security Class		ructural Base	as of Volt		~ Mo.	dal 0	hannala	
(0)	by Using Per	rfectly A	ligned Multi	es of vort laver Samp	age-gatin	.g Moi	der C	namers	
12. PERS	SONAL AUTHOR(S)							-	
13a TYI	Huang, Huey	13b. TIME CO	OVERED I	14. DATE OF REPO	RT (Year Mooth	Davi II	S. PAGE (OUNT	
	Annual	FROM 10	<u>/87</u> то <u>9/88</u>	9/20/88			4		
16 SUPI	PLEMENTARY NOTATION	i							
17	COSATI CO		18. SUBJECT TERMS (
FIEI		SU8-GROUP	Alamethicin Dependent C	; Melittin ircular Di	, Gramici Chroism.	din, X-ray	Orie V Dif	ntation fraction:	
	18		Alamethicin; Melittin; Gramicidin; Orientation Dependent Circular Dichroism; X-ray Diffraction; Neutron Scattering; Perfectly Aligned Multilayers						
	TRACT (Continue on rev		and identify by block n	umber)		-			
One-	dimensional	quasi-cry	stals of per	fect multi	layers, i	n wh	ich i	on chan-	
	are uniforms structural ba								
meth	techniques for preparing such multilayer samples and 2) the spectroscopic methods (circular dichroism and x-ray diffraction) for extracting struc-								
tural information from these samples. The sample variables include elec-									
tric field, water content, ion concentrations, etc. We have observed con-									
formation changes of alamethicin with water content, a result in favor of the barrel model (rather than the flip-flop model) for the channel. Our									
goal is to probe the conformation changes of the channels as we vary the									
sample variables, in order to elucidate the molecular mechanisms of									
	age-gating.	•	5						
	TRIBUTION / AVAILABILIT			21. ABSTRACT SEC	CURITY CLASSIFIC	ATION			
	☑ UNCLASSIFIED/UNLIMITED ☐ SAME AS RPT ☐ DTIC USERS			(U) 22b. TELEPHONE (Include Area Code) 22c. OFFICE SYM				M801	
	. Igor Vodya			202-696-		, , , , , , ,	ONR		
DD Form 1473 JUN 86 Previous editions are obsolete. SECURITY CLASSIFICATION OF THIS PAGE									
TP UBUT	ION STATEMENT A	.]							

Approved for public release: 88 9 26 90

Introduction

Because of the difficulty in making single crystals of membrane ion channels in their native forms (suitable for x-ray diffraction), there is a lack of structural information for understanding their molecular mechanisms. We believe that, under the circumstances, one-dimensional (1D) quasi-crystals of perfect multilayers, in which channels are uniformly oriented within parallel membranes, can be used to provide some of the much needed structural data. In the past few years, we have developed 1) the techniques for preparing such multilayer samples and 2) the spectroscopic methods for extracting structural information from these samples. Our goal is to investigate the conformation changes occurring in the channels when they are subject to electric field or variations in chemical conditions, in order to elucidate the molecular mechanisms of voltage-gating. A sensible approach to this complicated problem is to study simple model channels such as alamethicin and melitin first. However, it is important to point out that our method is applicable to natural proteins; for example, we have applied our method to study cytochrome b from yeast complex III in another research project. In the following, we review our objectives and the progress we made last year.

Objectives

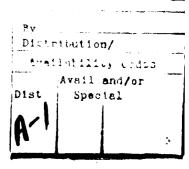
- 1. Preparing multilayer samples of gramicidin, alamethicin and melittin, and experimenting the variations of their chemical conditions.
- 2. Circular dichroism (CD) of multilayer samples to study the orientations of the α -helical sections in the channels.
- 3. X-ray scattering of multilayer samples to study the ion binding sites.
- 4. Normal incident neutron scattering of multilayer samples to study the channel distributions in membrane.
- 5. Electric field studies of multilayer samples to create the voltage-gating condition in the channels.

Accomplishments

Multilayer Samples -- We have by now successfully produced perfect multilayers of lipid-pepitde mixtures between two surfaces of fused silica, electrode (indium tin oxide) coated fused silica, mica and beryllium. Different substrata are used depending on the type of experiment. The thickness of multilayers can be varied between 1 and 100 µm. The sample variables include the peptide/lipid ratio, water content (15 to 40% of sample weight) and ion (e.g. Na⁺, K⁺, etc.) concentrations. The lipids used so far include dilauryl-, dimyristoyl-, dipalmitoyl-, and diphytanoyl-phosphatidylcholine (DLPC, DMPC, DPPC and DPhPC, respectively). These samples are free of smectic defects, transparent to light, and perfectly ordered in the direction perpendicular to the substrata surfaces (the mosaic spread ~ 0°) (for details see Huang and Olah, 1987 and Olah and Huang, 1988a).







- 2. CD -- Although a theory describing the dependence of CD on the orientation of an α-helix was known since the 1950's (Moffitt, 1956), experimentally it remained unproven until recently. The reasons are complicated and they are discussed in our recent papers (Olah and Huang, 1988a and 1988b). Because α-helices are somewhat flexible, only short (and hence straight) peptides can provide uniformly oriented α-helices. And in fact this condition has been achieved only in our perfectly aligned multilayer samples. We used the CD of alamethicin embedded in multilayers to prove the Moffitt theory and simultaneously established that the α-helical section of alamethicin was perpendicular to membrane under the condition we prepared the sample. A special technique was devised to measure CD of multilayers with light incident on the membranes at various tilted angles. Figure 1 shows an example of such measurement. The technique is very sensitive to the conformation changes of peptides. For example, Fig. 2 shows the changes of CD of alamethicin in DPhPC with hydration. As explained in the Fig. 2 caption, the results appear to be in favor of the barrel model (Hall, Vodyanoy, Balasubramamian and Marshall, 1984) rather than the dipole flip-flop model (Menestring, Voges, Jung and Boheim, 1986) for the formation of the alamethicin channel. A more complete analysis is in progress.
- 3. 1-D X-ray Diffraction -- Since our multilayers are in fact one-dimensional (1-D), quasi-crystals (perfect ordering in the direction normal to the planes of membranes), their 1-D electron density profiles can be determined by x-ray diffraction. It is, however, not easy to unravel the peptide signals from the intense diffraction background due to lipid bilayers. Therefore, our effort has been concentrated on measuring the ion distribution profiles, including the locations of the ion binding sites in the channels, by using heavy metal ions such as cesium and thallium. Early on we had difficulties with the x-ray absorption by the materials used to support the multilayers. This problem has now been solved by using beryllium. The diffraction of thallium ions in the gramicidin channels has been measured. Twelve Bragg peaks were recorded, amounting to 2-3 Å resolution. The analysis, including solving the phase problem, is in progress.
- 4. Normal Incident Neutron Scattering -- This experiment was designed to measure the 2-D protein distribution in the plane of membrane, which contains information about the protein-membrane interactions (Huang, 1986). Other research groups have attempted similar measurements of 2-D protein distributions by using vesicular samples and always found their signals masked by that of vesicles (despite their efforts of index matching with deuteration). On the contrary, the normal incident neutron scattering of defect-free, pure lipid multilayers gave a flat background (no angular dependence). Therefore if a protein has a reasonable neutron scattering contrast against the lipid background, its distribution can be measured. For this purpose, either peptide or lipid needs to be fully deuterated. We have placed an order with Avanti Polar Lipids (Pelham, AL) to synthesize fully deuterated DLPC. Their schedule and the problems in both the Oak Ridge and the Brookhaven neutron facilities have delayed the progress of this experiment.
- 5. Electric Field -- We have successfully coated indium tin oxide on fused silica surfaces so that an electric field of up to 50 kv can be applied across a multilayer sample (Olah and Huang, 1988b). The coated electrode is thin enough that it does not interfere with the CD measurement of the sample. However, the joule heating and the electrode damage at the anode have been problems. We have reduced the sample conductivities by purifying the chemical components. We have also coated the electrode with thin silicon dioxide to prolong its life. After many experiments, we found that we could apply to our samples electric field of square steps, each 0.1 s on and 0.9 s off, for hours. This allows at least two types of experiments. 1) Our CD spectrometer (Jasco J-500A) allows signal averaging with 0.1 s point measurements. Thus we can measure the CD of ion channels in electric field. 2) With synchrotron radiation, subnanosecond-resolved diffraction measurement is now possible (Science 241, 295, 1988). Thus we can measure the x-ray diffraction of ion channels in electric field.

Publications

- H. W. Huang, "Deformation Free Energy of Bilayer Membrane and Its Effect on Gramicidin Channel Lifetime" Biophys. J. <u>50</u>, 1061-1071 (1986).
- T. Y. Teng and H. W. Huang, "Hemoglobin and Myoglobin Embedded in Dry Polyvinyl Alcohol Film for X-ray Absorption Studies" Biochem. Biophys. Acta <u>874</u>, 13-18 (1986).
- H. W. Huang and G. A. Olah, "Uniformly Oriented Gramicidin Channels Embedded in Thick Monodomain Lecithin Multilayers" Biophys. J. <u>51</u>, 989-992 (1987).
- T. Y. Teng, H. W. Huang and G. A. Olah, "5K EXAFS and 40K 10-Second Resolved EXAFS Studies of Photolyzed Carboxymyoglobin" Biochemistry 26, 8066-8072 (1987).
- H. W. Huang, Book Review: "Accuracy in Molecular Processes" by T. B. L. Kirkwood, R. F. Rosenberger and D. J. Galas. Am. Sci. 76, 303 (1988).
- G. A. Olah and H. W. Huang, "Circular Dichroism of Oriented α-Helices I. Proof of the Exciton Theory" J. Chem. Phys. 89, 2531-2538 (1988).
- G. A. Olah and H. W. Huang, "Circular Dichroism of Oriented α-Helices II. Electric Field Oriented Polypeptides" J. Chem. Phys. Dec. (1988).



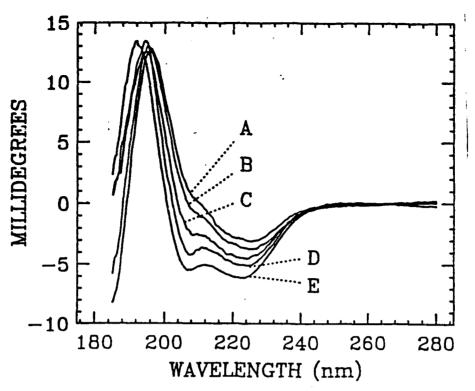


Fig. 1: Circular dichroism of oriented alamethic nembedded in DLPC multilayers at tilt angle (between the direction of light and the normal to the planes of bilayers) $\alpha = 0^{\circ}(A)$, 15°(B), 30°(C) and 45°(D). The amplitude at 208nm is proportional to $\sin^2\alpha$ as predicted by the Moffitt theory of α -helices; this is the first proof of the thoery (Olah and Huang, 1988a).

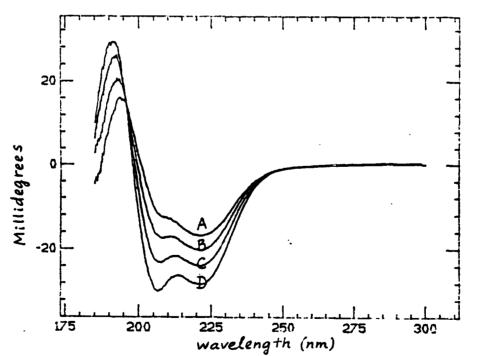


Fig. 2.: The changes of the circular dichroism of alamethicin embedded in DPhPC multilayers with the degree of hydration. Spectra were taken at normal incidence with the same sample by varying the equilibrating humidity: (A) 100% humidity; (D) 0% humidity; (B) and (C) in between. The results indicate that the insertion of α-helices into the membrane requires excessive water. This appears to be in favor of the barrel model (rather than dipole flip-flop model) for the formation of the alamethicin channel.

DISTRIBUTION LIST FOR REPORTS

ONR MEMBRANE ELECTROCHEMISTRY PROGRAM

Dr. Martin Blank
Department of Physiology
Columbia University College
of Physicians and Surgeons
630 W. 168th Street
New York, NY 10032

Dr. William E. Brownell
Department of Otolaryngology-HNS
Johns Hopkins University
School of Medicine
720 Rutland Avenue
Baltimore, MD 21205

Dr. Marco Colombini Department of Zoology University of Maryland College Park, MD 20742

Dr. Michael A. Cusanovich Department of Biochemistry University of Arizona Tuscon, AZ 85721

Dr. D. W. Deamer Department of Zoology University of California Davis, CA 95616

Dr. Edward A. Dratz Department of Chemistry Montana State University Bozeman, MT 59717

Dr. Harvey M. Fishman
Department of Physiology and
Biophysics
University of Texas Medical Branch
Galveston, TX 77550

Dr. Sol M. Gruner
Department of Physics
Jadwin Hall
Princeton University
P. O. Box 708
Princeton. NJ 08544

Dr. Felix T. Hong Department of Physiology Wayne State University 540 E. Canfield Avenue Detroit, MI 48201 Dr. Huey W. Huang Department of Physics Rice University Houston, TX 77251

Dr. Israel R. Miller Department of Membrane Research The Weizmann Institute of Science Rehovot 76100 ISRAEL

Dr. V. Adrian Parsegian Laboratory of Chemical Biology, NIADDK Room 9N-307 Building 10 Bethesda, MD 20892

Dr. Davis S. Perlin
Department of Biochemistry
Public Health Research Institute
455 First Avenue
New York, NY 10016

Dr. H. Gilbert Smith EG & G Mason Research Institute 57 Union Street Worcester, MA 01608

Dr. Michael E. Starzak Department of Chemistry State University of New York Binghamton, NY 13901

Dr. H. Ti Tien
Department of Physiology
Membrane Biophysics Laboratory
Michigan State University
East Lansing, MI 48824

Dr. Tian Y. Tsong
Department of Biological Chemistry
Johns Hopkins University
School of Medicine
725 N. Wolfe Street
Baltimore, MD 21205

Dr. Peter Vanysek Department of Chemistry Northern Illinois University De Kalb, IL 60115

ONR MEMBRANE ELECTROCHEMISTRY PROGRAM

Dr. Howard Wachtel
Dept. of Electrical & Computer Eng.
University of Colorado
Campus Box 425
Boulder, CC 80309

Dr. James C. Weaver Div. Health Sciences & Technology Room 20A-128 Massachusetts Institute of Tech. Cambridge, MA 20742

Dr. George S. Wilson Department of Chemistry University of Kansas Lawrence, KS 66045

Annual Final and Technical Reports

ADMINISTRATORS

Or. Igor Vodyanoy, Code 1141SB (2.copies) Scientific Officer, Biophysics Office of Naval Research 800 N. Quincy Street Arlington, VA 22217-5000

Administrator (2 copies) (Enclose DTIC Form 50)
Defense Technical Information Center
Building 5, Cameron Station
Alexandria, VA 22314

Administrative Contracting Officer ONR Resident Representative (address varies - obtain from contract or your business office) Dr. Robert J. Nowak, Code 1113ES Scientific Officer, Electrochemical Office of Naval Research 800 N. Quincy Street Arlington, VA 22217-5000

Program Manager Biological/Human Factors Division Code 125 Office of Naval Research 800 N. Quincy Street Arlington, VA 22217-5000

Program Manager Defense Technical Support Technology Directorate Office of Naval Technology, Code 223 800 N. Quincy Street Arlington, VA 22217-5000

Annual and Final Reports Only (one copy each)

DoD ACTIVITIES

Commander
Chemical and Biological Sciences Division
Research Army Research Office, P. O. Box 1221
Research Triangle Park, NC 27709

Head Biomolecular Engineering Branch Code 6190 Naval Research Laboratory Washington, DC 20375

Final and Technical Reports Only

Director, Naval Research Laboratory (6 copies) Attn: Technical Information Division, Code 2627 Washington, DC 20375 Directorate of Life Sciences Air Force Office of Scientific Bolling Air Force Base Research Washington, DC 20332